Betmip (1-(Triphenylphosphorylideneaminomethyl)benzotriazole), A Unique CH₂=N–PPh₃⁺X⁻ Equivalent for Organic Synthesis

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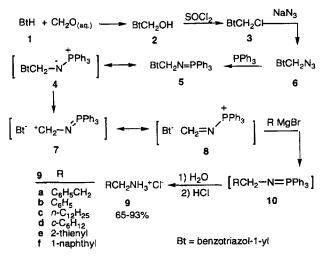
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Since the first syntheses of iminophosphoranes from tertiary phosphines and organic azides [1], numerous articles have described the synthetic applications of such N=P compounds [2–4]. We discuss betmip 5, 1-(triphenylphosphorylidene-aminomethyl)benzotriazole, a $^+CH_2N=PPh_3$ or $CH_2=N-P^+Ph_3$ synthon equivalent (an analog of vinyltriphenylphosphonium halide, $CH_2=CH^+PPh_3X^-$) for the generation of *N*-alkyliminophosphoranes. Betmip 5 is prepared from inexpensive aqueous formaldehyde and benzotriazole in four simple steps (Scheme 1) [5, 6], each affording a stable solid product in quantitative yield. The name betmip 5 consists of "bet", "m", "i", and "p" which represent "benzotriazole", "methylene", "imino", and "phosphine", respectively.

Betmip **5** contains both an electrophilic methylene carbon and a nucleophilic nitrogen atom. Nucleophilic substitution on the methylene carbon displaces benzotriazole with formation of a nucleophile-carbon bond, while reaction of an electrophile at the N=P group forms a nitrogen-electrophile bond. The nucleophilic and electrophilic centers of **5** mutually interact

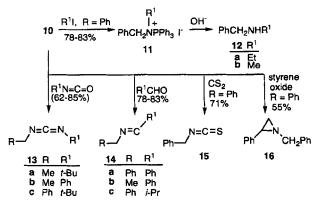




allowing easy formation of the ion pairs 7 and 8 as shown in Scheme 1, a lower reactivity of N=P towards electrophiles than for other iminophosphoranes, and easier displacement of the benzotriazole by nucleophiles as compared to other benzotriazole Mannich derivatives. Thus betmip 5 has good stability, allowing prolonged storage.

Grignard reagents transform betmip **5** into iminophosphoranes **10** [5] which undergo hydrolysis during work-up to give amines. Thus, betmip **5** provides a convenient ${}^{+}CH_2NH_2$ synthon equivalent for conversion of a carbanion to an one-carbon higher homologous primary amine. *N*,*N*-Bis(trimethylsilyl)methoxymethylamine, the only similar synthon equivalent [7, 8], is prepared from hexamethyldisilane and carcinogenic chloromethyl methyl ether. Product yields are significantly higher by the betmip method than using *N*,*N*-bis(trimethylsilyl)methoxymethylamine.

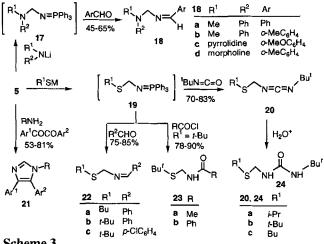
In situ treatment of the same intermediates 10 with allkyl iodides gave 11, easily hydrolyzed to secondary amines 12. Similarly, 10 with an isocyanate, an aldehyde, carbon disulfide, or styrene oxide afforded in high yield unsymmetrical carbodiimides 13, Schiff bases 14, isothiocyanates 15, and aziridines 16 (Scheme 2) [6]. These reactions can be carried out in one-pot from 5, without isolation of iminophosphoranes 10, with straightforward work-up. The by-prod-





uct benzotriazolate is easily removed during aqueous workup and could be recovered for reuse.

N-(α -Heteroatomalkyl)-substituted iminophosphoranes had not been reported prior to the discovery of betmip 5. Treatment of betmip 5 with lithium amides formed N-(α -dialkylaminomethyl)iminophosphoranes 17, which were converted by arylaldehydes to novel α -(arylideneamino)alkylamines 18 (Scheme 3) [9]. Similarly, N-(α -alkylthiomethyl)iminophosphoranes 19, conveniently prepared by reaction of betmip 5 with thiolates, were transformed without isolation into four novel classes of functionalized formaldehyde-N,S-acetals [10]; N-(alkylthiomethyl)-N'-alkylcarbodiimides 20, N-(alkylthiomethyl)benzalimines 22, N-(alkylthiomethyl)amides 23 and N-(alkylthiomethyl)-N'-alkylureas 24 (Scheme 3).

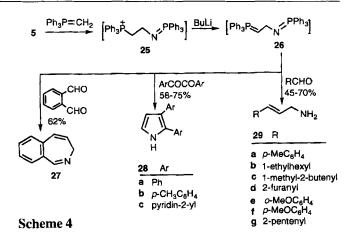




Primary amines and α -diaryl ketones convert betmip into 1.4.5-trisubstituted imidazoles 21 in good yield [11]: this is the first method to synthesize 1,4,5-trisubstituted imidazoles from α -diketones.

Bis(alkylidenetriphenylphosphoranes) (1,2-, 1,3-, and 1,4bisylides) and bis(iminophosphoranes) have been known for decades [2, 6, 12]. However, the chemistry of monoazabisphosphorus ylides containing both an iminophosphorane and a phosphorus ylide was unexplored prior to betmip chemistry. Nucleophilic reaction of betmip 5 with methylidenetriphenylphosphorane followed by treatment with butyl- lithium gave 1-aza-1,3-bis(triphenylphosphoranylidene) propane 26, the first 1,3-monoazabisylide (Scheme 4) [13, 14]. In situ treatment of the 1,3-monoazabisylide 26 with phthalic dicarboxaldehyde afforded 3H-2-benzazepine 27 in 62% yield. Wittig/aza-Wittig reactions of 26 with α -diketones gave 2,3disubstituted pyrroles 28 in 58-75% yield. The different reactivity of the two functionalities in 26 with aldehyde allowed the development of an efficient new synthetic method for primary allylamines 29. Thus, treatment of 26 with one equivalent of aldehydes gave primary allylamines 29 in high steric selectivity in 45-75% yields (Scheme 4).

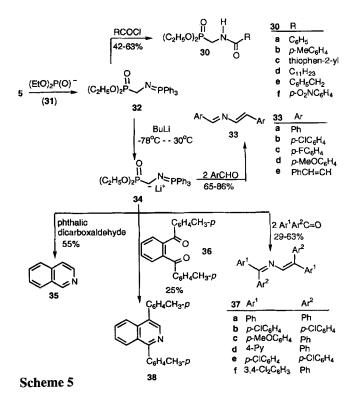
Similarly, nucleophilic displacement of benzotriazole from 5 with anion 31 (from dimethyl phosphite and butyllithium) gave intermediate 32, a precursor of 1,2-monoazabisylide equivalent (Scheme 5) [13, 15]. Intermediate 32 was converted



by acyl chlorides into diethyl [(acylamino)methyl]phosphonates 30.

The methylene group in 32 can also be deprotonated by butyllithium at -30 °C to generate phosphonate carbanion 34 (a 1,2-monoazabisylide equivalent). Treatment of 34 with two equivalents of an aldehyde gave 2-azabutadienes 33 (Scheme 5). However, intermediate 32 reacts with a single molar equivalent of an aldehyde to give the 1:2 product 33 because one equivalent of 32 is consumed in a Horner-Wittig step. Horner-Wittig/aza-Wittig reaction of 34 with diaryl ketones, phthalic dicarboxaldehyde or diketones afforded 1,1,4,4-tetraarylbutadienes 37 and isoquinolines 35, 38, respectively.

In conclusion, betmip as the first CH₂=N-P+Ph₃ X⁻ aza-Wittig analog provides a new carbanion route to iminophosphoranes and convenient synthon equivalents of +CH₂NH₂ and ⁺CH₂N=. Betmip also allows for the first preparation of



new α -heteroatom-alkyl iminophosphoranes and 1,2-monoazabisylide equivalents. The synthetic utility of betmip has been illustrated by the convenient synthesis of a broad spectrum of nitrogen-containing compounds and heterocycles.

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